Biomaterials in Implants

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Historical background

- Tooth loss from disease has always been a feature of mankind's existence.
- For centuries people have attempted to replace missing teeth using implantation.
- Greek, Egyptian civilization used materials like jade, bone, carved ivory, metal and even animal teeth

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- Since then many types of implant materials were introduced but consistent failures occured with them
- In 1952 Branemark developed a threaded implant design made of pure titanium that showed direct contact with bone.
- Henceforth popularity of implants reached new heights

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Introduction

- Currently the implant materials available are diverse
- Success and longevity of implant depends on fours B's
 - Biomaterial
 - Biomechanics
 - Biological tissues
 - Body serviceability
- Definitive need to have a knowledge of these biomaterials for their judicious selection and application in implantology.

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Definition

Biomaterials

- According to GPT-8
- Any substance other than a drug that can be used for any period of time as part of a system that treats ,augments or replaces, any tissue ,organ or function of the body

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Classification

According to composition

- 1. Metal and Metal alloys
 - Titanium
 - Titanium alloys (Ti6Al4V)
 - Cobalt, Chromium, Molybdenum alloy (Vitallium)
 - Austenitic steel or Surgical steel (Iron, Chromium, Nickel alloy)
 - Precious metals (Gold , Platinum, Palladium)

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According to composition

2. Ceramics and carbon

- Aluminium oxide Alumina
 - Sapphire
 - Zirconium oxide (zirconia)
 - Titanium oxide (titania)
- Calcium phosphate ceramics (CPC)
 - Hydroxyapatite (HA)
 - Tricalcium phosphate(TCP)
 - Glass ceramics
 - Vitreous carbon (C), Carbon-silicon(C-Si)

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According to composition

3. Polymers

- Poly methyl metha acrylate (PMMA)
- Poly tetra fluoro ethylene (PTFE)
- Poly ethylene terapthylate (Dacron)
- Dimethyl polysiloxane(Silicone rubber)
- Ultrahigh molecular weight polyethylene(UHMW PE)
 - Poly sulphone
- 4. Composites (combination of polymer and other synthetic biomaterials)
 - Carbon PTFE
 - Carbon- PMMA
 - Alumina- PTFE

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Biological classification – according to tissue response 1. Biotolerant – Polymers Fibrous tissue encapsulation at implant interface 2. Bioinert – Titanium, Ti alloy, Alumina, zirconia Direct bone apposition at the implant interface 3. Bioactive- CPC, Glass ceramics Direct chemical bonding of implant with the surrounding bone

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Requirements for an ideal implant material

- Any material intended for use as dental implant must meet 2 basic criteria
 - Biocompatibility with living tissue
 - Biofunctionality with regard to force transfer
- Implant material should have certain ideal physical, mechanical, chemical and biological properties to fulfill these basic criteria
- Implant properties can be studied under
 - Bulk properties
 - Surface properties

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Bulk properties

Modulus of elasticity (E)

- Measure of change in dimension (strain) with respect to stress
- Ideally a biomaterial with elastic modulus comparable to bone (18GPa)should be selected
- This will ensure more uniform distribution of stress at implant bone interface as under stress both of them will deform similarly.
- Hence the relative movement at implant bone interface is minimized.

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Tensile, Compressive, Shear, Strength

- Forces exerted on implant material consists of tensile, compressive, shear components
- An implant material should have high tensile, compressive, shear strength to prevent fractures and improve functional stability.

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Yield strength and Fatigue strength

- Yield strength is magnitude of stress at which a material shows initial permanent deformation
- Fatigue strength is stress at which material fractures under repeated loading
- An implant material should have high yield strength and fatigue strength to prevent brittle fracture under cyclic loading

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Ductility

- Refers to relative ability of a material to deform plastically under a tensile stress before it fractures
- ADA demands a minimum ductility of 8% for dental implant
- Required for fabrication of optimal implant configurations
- Safeguards against brittle fractures of implant

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Hardness and Toughness

- Hardness resistance to permanent surface indentation or penetration
- Increase hardness decreases the incidence of wear of implant material
- Toughness amount of energy required to cause fracture.
- Increased toughness prevents fracture of the implants.

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Electrical and Thermal conductivity

 Should be minimum to prevent thermal expansion, contraction, and oral galvanism.

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Surface properties

Surface tension and surface energy

Determines

- Wettability of implant by wetting fluid (blood)
- Cleanliness of implant surface
- Surface energy of > 40 dyne / cm
- Surface tension of 40 dyne/cm or more
- Characterstics of very clean surface
- Results in good tissue integration with load carrying capacity

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Biocompatibility

- Not total inertness
- Ability of a material to perform with an appropriate biological response in a specific application
- Mainly a surface phenomenon
- Most important requirement for a biomaterial
- Depends on
 - Corrosion resistance
 - Cytotoxicity of corrosion products

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Corrosion resistance

- Corrosion is deterioration of a metal caused by reaction with its environment
- Following types of corrosion are seen
 Stress corrosion
- Failure of a metal by cracking due to increased stress
- **Fretting corrosion**
- Due to micromotion or rubbing contact within a corrosive environment

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Crevice corrosion

 Occurs in narrow region eg implant screw – bone interface

Pitting corrosion

- Occurs in surface pit
- Metal ions dissolve and combine with CI ions

Galvanic corrosion

 Occurs between two dissimilar metal in contact within an electrolyte resulting in current flow between the two.

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Electrochemical corrosion

- In this anodic oxidation and cathodic reduction takes place resulting in metal deterioration as well as charge transfer via electrons.
- All these types of corrosion and charge transfer can be prevented by presence of passive oxide layer on metal surface.
- The inertness of this oxide layers imparts biocompatibility to biomaterials

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Cytotoxicity of corrosion products

- Toxicity of implant materials depends on toxicity of corrosion products which depends on
 - Amount of material dissolved by corrosion per unit time
 - Amount of corroded material removed by metabolic activity in same unit time
 - Amount of corrosion particles deposited in the tissue
- Both increased corrosion resistance and decreased toxicity of corrosion products contribute to biocompatibility

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Bone and implant surface interaction

- The implant material should have an ability to form direct contact or interaction with bone (osseointergration)
- This is largely dependent on biocompatibility and surface composition of biomaterial (presence of passivating oxide layer).

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Biomaterials

Titanium

- Gold standard in implant materials
- Composition of Commercially pure titanium
 - Titanium 99.75%
 - Iron 0.05%
 - Oxygen 0.1%
 - Nitrogen 0.03%
 - Hydrogen 0.012%
 - Carbon 0.05%
- Commercially pure titanium occurs in 4 grades, grade I II III IV, according to oxygen content (0.18% to 0.40%) & iron content (0.20 to 0.50 wt %)

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- Consists of 2 phases α and β phase
- Ti implants are mainly manufactured through controlled machining (lathing, threading, milling)
- Configuration like cylinders ,screws and blade forms etc are used
- Casting of titanium alloy is difficult due to high melting points (1700°C)
- Also titanium readily absorbs nitrogen hydrogen and oxygen from air during casting which makes it brittle

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Properties

Biocompatibility

- Titanium is one of the most biocompatible material due to its excellent corrosion resistance
- The corrosion resistance is due to formation of biologically inert oxide layer

Oxide layer

- Titanium spontaneously forms tenacious surface oxide on exposure to the air or physiologic saline
- Three different oxides are
 - TiO Anastase
 - TiO₂ Rutile
 - Ti₂O₃ Brookite

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- TiO₂ is the most stable and mostly formed on titanium surface
- This oxide layers is self healing i.e. if surface is scratched or abraded during implant placement it repassivates instantaneously
- Also Ti oxide layer inhibits low level of charge transfer, lowest among all metals. This is the main reason for its excellent biocompatibility

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- Good yield strength , tensile strength , fatigue strength .
- Modulus of elasticity (110 GPa) is half of other alloys and 5 times greater than bone
 This helps in uniform stress distribution
- Good strength ,but less than Ti alloys.
- Ductile enough to be shaped into implant by machining
- Low density 4.5g/cm³, light weight

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- Titanium allows bone growth directly adjacent to oxide surface
- Inspite of excellent corrosion resistance peri-implant accumulation and also accumulation in lung, liver, spleen of Ti ions is seen, however in trace amount it is not harmful
- Increased level of titanium ions can result in titanium metallosis.

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Titanium alloys Ti6Al4VConsists of

- Titanium
- -6% Aluminium alpha stabilizer
- -4% Vanadium beta stabilizer



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Properties

- Excellent corrosion resistance
- Oxide layer formed is resistant to charge transfer thus contributing to biocompatibility
- Modulus of elasticity is 5.6 times that of the bone ,more uniform distribution of stress
- Strength of titanium alloy is greater than pure titanium – 6 times that of bone hence thinner sections can be made

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- Ductility is sufficient
- Exhibits osseointergration

Uses

 Extensively used as implant material due to excellent biocompatibility ,strength ,osseointegration

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Cobalt, Chromium, Molybdenum alloy

- Composed of same elements as vitallium
- Vitallium introduced in 1937 by Venable Strock and Beach
- Composition
 - -63% Cobalt
 - 30% Chromium (CrO provides corrosion resistance)
 - 5% Molybdenum(strength)

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Properties

- High mechanical strength
- Good corrosion resistance
- Low ductility
- Direct apposition of bone to implant though seen ,it is interspersed with fibrous tissue

Uses

 Limited for fabrication of custom designs for subperiosteal frames due to ease of castability and low cost.

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Iron, Chromium, Nickel based alloy

- These are Surgical steel alloys or Austenitic steel
- Have a long history of use as orthopedic and dental implant devices

Composition

- Iron
- Chromium 18% corrosion resistance
- Nickel 8% stabilize austenitic steel

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Properties

- High mechanical strength
- High ductility
- Pitting and crevice corrosion.
- Hypersensitivity to nickel has been seen
- Bone implant interface shows fibrous encapsulation and ongoing foreign body reactions
- Use is limited

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Precious metals

- Gold , Platinum , Palladium
- They are noble metals unaffected by air, moisture, heat and most solvents
- Do not depend on surface oxides for their inertness
- Low mechanical strength
- Very high ductility
- More cost per unit weight
- Do not demonstrate osseointegration
- Not used

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Ceramics

- Ceramics are inorganic, non metallic materials manufactured by compacting and sintering at elevated temperature
- Consist of
- Bioinert ceramics
 - Aluminium oxide
 - Titanium oxide
 - Zirconium oxide
- Bioactive ceramics
 - Calcium phosphate ceramics (CPC) hydroxyapatite (HA) tricalcium phosphate (TCP)
 - Glass ceramics

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Bioinert Ceramics

 These ceramics show direct bone apposition at implant surface but do not show chemical bonding to bone

Properties

- Bioinert ceramics are full oxides i.e. bulk and surface thus excellent bio compatibility
- Good mechanical strength
- Low ductility which results in brittleneSS
- Color similar to hard tissue

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Uses

- Though initially thought to be suitable for load bearing dental implants but to due inferior mechanical properties
- Used as surface coatings over metals
 - to enhance their biocompatibility
 - to increase the surface area for stronger bone to implant interface

Bioactive ceramics

Calcium phosphate ceramics

- These ceramics have evoked greatest interest in present times
- Mainly consists of
 - Hydroxyapatite(HA)
 - Tricalcium phosphate(TCP)

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Properties

Biocompatibility

- CPC have biochemical composition similar to natural bone
- CPC form direct chemical bonding with surrounding bone due to presence of free calcium and phosphate compounds as implant surface
- Excellent biocompatibility
- No local or systemic toxicity
- No alteration to natural mineralization process of bone

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- Lower mechanical tensile and shear strength
- Lower fatigue strength
- Brittle, low ductility
- Exists in amorphous or crystalline form
- Exists in dense or porous form
 - Macro porous > 50 μm
 - Micro porous < 50 μ m

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The pores though decrease the strength they increase the surface area providing additional region for tissue ingrowth
Ideal pore size is around 150µm, same diameter as shown by inter trabecular spaces in bone

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Solubility of CPC

- CPC show varied degree of resorption or solubility in physiologic fluids
- The resoption depends on
 - Crystallinity
 - High crystallinity is more resistant to resorption
 - Particle size
 - Large particles size requires longer time to resorb

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Porosity

Greater the porosity, more rapid is the resorption.

Local environment

 Resorption is more at low pH eg in case of infection or inflammation

Purity

presence of impurities accelerate resorption

It has been seen that HA resorb less readily than TCP

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Uses

- Due to lack of mechanical strength, not used as load bearing implants
- Used as Bone grafts material for augmentation of bone
- As bioactive surface coating for various implant material to increase
 - biocompatibility
 - strength of tissue integration

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Glass ceramics

- They are bioactive ceramics
- Bioglass or Ceravital
- Silica based glass with additions of calcium and phosphate produced by controlled crystallization

Properties

- High mechanical strength
- Less resistant to tensile and bending stresses
- Extremely brittle

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 They chemically bond to the bone due to formation of calcium phosphate surface layer

Uses

- Inferior mechanical properties not used as load bearing implant
- Used more often as bone graft material
- When used as coating bond between coating and metal substrates is weak and subject to dissolution

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Carbon and carbon silicon compounds

 Vitreous Carbon and Carbon compounds (SiC)were introduced in 1960 for use in implantology

Properties

- Inert
- Biocompatible
- Modulus of elasticity is close to that of bone
- Bone implant interface shows osseointegration
- Brittle
- Susceptible to fracture under tensile stress
 Uses
- Used mainly as surface coatings for implants materials

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Polymers

- Polymeric implants were first introduced in 1930s
- However they have not found extensive use in implant due to
- Low mechanical strength
- Lack of osseointegration

- Used currently to provide shock absorbing qualities in load bearing metallic implants.
 E.g. in IMZ system a polyoxymethylene intra mobile element (IME) is placed between prosthesis and implant body which
 - Ensures more uniform stress distribution
 - Acts as internal shock absorber

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Composites

- Combination of polymer and other synthetic biomaterial.
- They have advantages that properties can be altered to suit clinical application
- Have a promising future .

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Surface characterization and preparation of implants

- Interaction between host tissue and implant primarily occurs at implant surface
- Thus characterization of implant surface to suit clinical needs is of paramount importance
- Surface characterization can be accomplished by following techniques

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Passivation –

- Refers to enhancement and stabilization of oxide layer to prevent corrosion
- Performed by immersion in 40% nitric acid
- Used for Co Cr implant

Acid etching

- In this the surface is treated with nitric or hydrofluoric acid
- Results in clean surface with roughened texture for increased tissue adhesion

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Sand blasting

- Sand particles are used to get a roughened surface texture which
 - increases the surface area
 - increases the attachment strength at the bone implant interface

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Surface coatings

- Implant surface may be covered with porous coatings which increases
- Surface area and roughness
- Attachment strength at bone implant interface
- Biocompatibility
- Several coating techniques exist.
- Plasma sprayed technique is used most commonly
- Two types
 - Plasma sprayed titanium
 - Plasma sprayed hydroxyapatite

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Plasma sprayed Titanium

- Described by Schroeder et al (1976)
- Titanium particles with mean size of 0.05 to 0.1 are heated in plasma flame
- Plasma flame consists of electric arc through which argon gas stream passes
- A magnetic coil directs the stream of molten titanium particles, which is then sprayed on the titanium surface

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- Thickness of coating 0.04 to 0.05 mm
- Sprayed coating exhibits round pores that are interconnected ,pore diameter(150-400µm)



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Advantages

- Increases the surface area by 600%
- Increases attachment of implant to bone
- Increase load bearing capacity

Disadvantages

- Cracking and exfoliation of the coating due to stresses ,sterilization and insertion.
- Metallic particles found in perimplant tissue

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Plasma sprayed hydroxyapatite

- Herman 1988
- Crystalline HA powder is heated to a temperature of 12000 to 16000 °C in a plasma flame formed by a electric arc through which an argon gas stream passes.
- HA particle size is approximately 0.04mm
- The particles melt and are sprayed on to the substrate ,they fall as drops and solidify
- Round interconnected pores are formed

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- Coating bonds to substrate by mechanical interlocking
- Coating of 0.05 mm is formed
- There is a lot of controversy regarding the ideal coating thickness
- Studies have shown that:
- Fracture occurred in coatings more than 0.1mm in thickness
- Whereas bioresorption was unacceptably rapid with coatings less than 0.03mm in diameter
- Ideal coating thickness of 0.05 mm is recommended

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Advantages

- Permits direct chemical bonding of the bone to implant surface
- Increases surface area
- Stronger bone to implant interface
- Increases corrosion resistance and biocompatibility
- Decreases healing period of implants
- Bone adjacent to coated implant is better organized and mineralized ,thus increased load bearing capacity

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Disadvantages

- Studies have shown that coatings exhibit several drawbacks:
- Coatings have shown to undergo gradual resorption over time and subsequent replacement with bone (creeping substitution)
- Studies have shown that this resorption results in decreased % of bone implant contact area over time
- Due to resorption of the coating ,biocompatibility of exposed and altered core substrate becomes questionable

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- Bond strength between coating and the substrate seems to be inadequate to resist shear stresses
- Due to this weak bond ,coating is susceptible to removal or fracture during
 - sterilization
 - insertion in dense bone
- Due to roughened surface, coating often shows adherence of microorganisms on their surface

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Indications

- HA coated implants can be used in
- D3 and D4 bone which show poor bone density and structure as they
 - Increase bone contact levels
 - Forms stronger bone implant interface
 - Increases survival rates
- Fresh extraction sites as they promote
 - Faster healing
 - Greater initial stability
- Newly grafted sites where implants are to be placed eg sinus lifts

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Alternative surface coating techniques

Electrophoretic deposition

- Mineral ions that need to be coated on the implant surface are dissolved in the electrolytic bath
- Current flows through electrolyte leading to formation of surface coating

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Sol gel deposition (Dip Coating)

 Coating is applied on substrate by dipping into a solution HA powder and ethanol in dip coating apparatus and finally sintering

it

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Hot isostatic pressing

- In this the HA powder is mixed with water and sprayed on the substrate
- It is then hot pressed at 850°C
- The coating produced is dense having increased shear strength

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Pulsed laser deposition

- Alternative procedure to obtain HA coating
- Nd YAG laser beam is used to spray HA on the preheated substrate in a vacuum chamber
- HA coating of greater crystallinity is obtained that shows decreased resorption

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Sterilization

- Today in most cases manufacturers guarantees precleaned and presterilized implants ,ready to be inserted
- In case the implants needs to be resterilized conventional sterilization techniques are not satisfactory
- Steam sterilization
 - should not be used as it results in contamination of surfaces with organic substances
- Dry heat sterilization
 - Also leaves organic and inorganic surface residues

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Radio frequency glow discharge technique (RFGDT) or Plasma cleaning

- Most frequently used methods
- In this, material to be cleaned is bombarded by high energetic ions formed in gas plasma in a vacuum chamber
- Removes both organic and inorganic contaminants

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UV light sterilization

- Recently UV light sterilization is also being used
- It cleans the surface and also increase the surface energy

Gamma radiation

- Method used to sterilize pre packaged dental implants.
- Radiation dose exceeding 2.5 megavolts is given
- Components remain protected, clean and sterile until packaging is opened, within sterile field of surgical procedure

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Conclusion

- A wide range of biomaterials are currently in use
- Appropriate selection of biomaterials directly influences, clinical success and longevity of implants
- Thus the clinician needs to have adequate knowledge of the various biomaterials and their properties for their judicious selection and application in his clinical practice

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